

# The Crystal and Molecular Structure of Racemic Potassium Di- $\mu$ -tartrato-diantimonate (III) Trihydrate (Racemic 'Tartar Emetic')

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Three-dimensional X-ray analysis shows that racemic potassium antimonyl tartrate is the racemic mixture of potassium di- $\mu$ -(+)-tartrato-diantimonate(III) and di- $\mu$ -(-)-tartrato-diantimonate(III) with three water molecules of crystallization. There are four formula units of  $K_2[Sb_2(C_4H_2O_6)_2] \cdot 3H_2O$  in the orthorhombic unit cell of dimensions  $a = 8.79$ ,  $b = 16.32$ ,  $c = 12.19$  Å and space group  $Pca2_1$ . Two antimony atoms are bridged by two tartrate groups which act as double bidentate ligands through the deprotonated carboxyl group and  $\alpha$ -hydroxyl oxygen atoms with mean Sb-O distances of 2.20 and 2.01 Å respectively. The chelate rings are nearly planar and the oxygen atoms about the antimony atom occupy four corners of a slightly distorted square pyramid, whose apex belongs to the unshared electron pair. Two of the three molecules of water per formula unit form a three-dimensional hydrogen-bonded network with tartrato-antimonate ions, while the third water molecule forms separate hydrogen bonds with the anion.

## Introduction

The crystal structures of four compounds considered to be 'antimonyl-tartrates' have been reported: ammonium antimonyl-(±)-tartrate monohydrate (Kiosse, Golovastikov & Belov, 1964); potassium antimonyl-(±)-tartrate hemihydrate (Grdenić & Kamenar, 1965), tris-*o*-phenanthroline-iron(II) antimony(+) tartrate (Templeton, Zalkin & Ueki, 1966); ammonium antimony-(+)-tartrate hemihydrate (Kiosse, Golovastikov, Ablov & Belov, 1967). Essentially the same dimeric tartratoantimonate(III) ion  $[Sb_2(C_4H_2O_6)_2]^{2-}$  was found in the crystal structures of ammonium compounds and the complex iron compound, but in the potassium compound a monomeric ion was reported. The latter result was obtained on the basis of only one projection and the structure was not completed at the time. The three-dimensional structure analysis, the results of which we now report, has shown that the dimeric tartratoantimonate ion occurs also in the crystal structure of racemic potassium antimonyl tartrate.

## Experimental

Racemic potassium antimonyl tartrate, known as hemihydrate, with the formula  $KSbOC_4H_4O_6 \cdot \frac{1}{2}H_2O$  was prepared by adding antimony(III) oxide to a warm solution of racemic tartaric acid and potassium hydroxide in water. The crystals obtained were recrystallized from an aqueous solution.

Cell dimensions, determined from oscillation and Weissenberg photographs, were as follows:

$$a = 8.79 \pm 0.02 \text{ \AA}$$

$$b = 16.32 \pm 0.02$$

$$c = 12.19 \pm 0.02$$

The systematic absences of  $0kl$  for  $l$  odd and  $h0l$  for  $h$  odd indicated  $Pca2_1$  ( $C_{2v}^5$ ) or  $Pbcm$  ( $D_{2h}^{11}$ ) as the possible space groups. The final choice of  $Pca2_1$  was ratified in the course of the Patterson synthesis interpretation and by piezoelectric measurements. The cell contains 8 formula units  $KSbOC_4H_4O_6 \cdot \frac{1}{2}H_2O$ ; measured density 2.510 g.cm<sup>-3</sup>, calculated density 2.508 g.cm<sup>-3</sup>. Three-dimensional intensity data were obtained up to the 10th layer around [001] and up to the 7th layer around [100] on an integrating Weissenberg camera using the multiple-film technique and Cu  $K\alpha$  radiation. For the photographs taken around [001] the crystal was spherically ground to 0.43 mm in diameter and for the data taken around [100] a sphere of 0.50 mm in diameter was used. The intensities were measured by means of a densitometer and corrected for Lorentz and polarization factors, and for absorption ( $\mu = 315 \text{ cm}^{-1}$ ), and were placed on a common scale by the method of Rollett & Sparks (1960).

Structure factors, cycles of least-squares analysis, and Fourier syntheses were calculated by the use of J.S. Rollett's general crystal structure analysis program on the Oxford University English Electric Leo-Marconi KDF. 9 computer.

## Structure determination

A three-dimensional Patterson function sharpened to 'point atoms at rest' computed from 1873 independent reflexions gave the positions of the antimony and potassium atoms. The light atom positions were found by means of several successive Fourier syntheses. The structure was then refined by several cycles of least-squares using the data uncorrected for absorption. At this stage of refinement the reliability index was  $R = 0.129$ . After absorption correction the refinement pro-

Table 1. Observed and calculated structure factors ( $\times 100$ ) with calculated phases

$h$	$k$	$l$	$F_o$	$F_c$	$\alpha_{hkl}$	$h$	$k$	$l$	$F_o$	$F_c$	$\alpha_{hkl}$	$h$	$k$	$l$	$F_o$	$F_c$	$\alpha_{hkl}$	$h$	$k$	$l$	$F_o$	$F_c$	$\alpha_{hkl}$		
0	0	2	6235	7098	93	0	15	10	4785	4728	177	1	7	7	7294	6323	-151	1	16	0	2906	1435	-180		
4		7125	7064	-3		0	16	0	4563	4231	0	8	10678	10271	-160	1	8264	7299	-46	2	5	0	8423	7040	-180
6		18682	22455	-25		2	7048	6811	146			9	10392	10094	-24	5	6694	5193	-24	1	11699	11680	82		
8		22640	26478			4	10562	11027	-180	10	5811	4931	-110		2886	2001	29	2	11282	10824	39				
10		8126	9312	116		6	5701	5197	-105	11	9610	9868	-35	4	4653	4001	29	3	11447	11423	172				
12		5885	7270	-17		8	2193	2472	165	12	5262	5152	13	5	4544	3898	-162	4	4505	4464	-35				
14		6238	7510	-1		0	17	0	8420	8170	180	13	5758	5816	165	6	4460	3706	145	5	12245	11675	4		
14		4839	9247	-25		4	7954	8135	1	1	14476	15077	180	8	4975	4783	-114	7	6638	5962	122				
14		10895	10119	88		6	5741	6007	127	1	8	14776	15077	180	8	2026	1952	-4	7	12174	12607	107			
8		5493	7334	18		8	2191	2519	-154	2	4749	5759	177	1	17	0	5655	4454	0	9	5152	4037	131		
12		7555	8993	-55	0	18	0	4675	5652	0	3	6759	6217	154	1	6475	6109	-66	10	6178	5732	-47			
12		4632	4456	17		2	5875	5344	5	4	15172	14700	-179	2	7541	7518	-23	11	5135	2588	170				
14		3980	4497	70		4	5073	4484	15	5	5726	5059	26	3	2082	1820	-54	12	5163	2710	70				
0	2	19595	16021	-138		6	7068	6177	-6	6	5634	4561	-164	4	5515	5012	25	13	5550	5419	21				
4		10794	10746	69	0	19	0	8700	10024	0	7	11886	11213	102	5	4060	3441	-104	2	6	0	21039	20290	0	
6		17179	11759	129	2	6284	6860	-41	8	6982	6707	178	6	4827	4581	7	1	15174	15229	54					
8		10559	11534	-139	4	1983	2075	-124	9	6555	5756	108	7	4869	4765	-111	2	11924	9808	145					
12		2815	2550	-25	0	20	0	5297	2562	180	10	5523	4224	-161	8	4052	3709	-17	9	21890	21389	85			
14		4773	5894	169	1	1	16703	20044	180	11	2040	1956	102	1	18	0	5297	5359	0	3	14054	15211	-165		
0	3	25549	21404	158	1	1	13943	14414	3	12	6581	7202	-170	1	5800	5272	145	3	14434	15601	112				
4		17587	18103	-152	1	1	14727	17308	-132	13	5275	5175	-157	3	7855	7621	-115	6	6178	5207	137				
6		17981	20353	-165	2	1	14922	17302	-132	1	8560	7162	190	4	2277	1617	-125	7	6178	5207	137				
8		19256	14868	129	3	6121	6075	17	1	15010	11857	123	5	5816	6156	-34	8	4024	2981	16					
10		12120	18109	158	5	15005	17519	135	2	4922	4112	88	6	3639	3625	-38	9	11101	10668	52					
12		10208	14516	-140	6	15005	17519	135	3	7570	6852	-156	7	6233	3991	69	10	6642	6099	152					
14		5435	2998	-167	6	7924	8627	178	4	4401	3750	169	1	19	0	2984	2518	180	11	10890	11955	94			
0	4	5409	6085	180	7	6303	7205	-167	5	7072	6047	-25	1	1359	1270	-10	12	5583	2889	-128					
2		5154	2963	169	8	9166	11307	-154	6	5915	5149	-118	2	5418	5619	-154	13	5796	6478	152					
4		8583	7990	55	9	8592	9809	53	7	9880	8972	80	3	4234	4425	87	14	1954	2349	-48					
6		10985	12293	164	10	5983	6387	-144	8	4934	4135	63	4	3694	2937	159	2	3647	2236	180					
8		9403	9800	-147	11	9880	9835	101	9	6294	5169	-180	5	2545	2078	154	7	8994	8867	111					
10		5791	5621	164	12	2731	2947	175	10	2906	2090	158	1	20	0	6048	7428	180	2	4440	3388	178			
12		5615	3452	161	13	5973	6095	170	11	2476	2302	-99	1	4151	3515	23	3	11261	10665	95					
14		4718	5205	171	14	6026	6803	175	12	3181	9233	-156	2	5070	5074	-179	5	9564	8527	90					
0	5	11297	11752	-74	1	2	0	7570	7567	-180	13	2689	2778	9	3	4575	5903	93	6	5591	2118	150			
4		19158	21486	50	1	14629	17396	108	14	0	5095	3728	0	2	0	1	8914	7144	57	7	8994	8867	111		
6		19222	10844	90	2	15889	20123	144	15	16539	15052	-58	2	1478	1676	93	8	5056	4154	58					
8		9058	9928	-97	3	4499	2056	-84	9	10740	9758	-40	3	26173	30793	87	9	5197	4246	80					
10		8680	9058	-8	4	11140	11682	-154	11	10741	9758	-40	4	15017	20029	105	10	2429	2436	-114					
12		8607	9454	52	5	5129	5262	49	12	5900	4856	65	5	15017	20029	105	11	5268	5258	102					
14		2855	3445	129	6	8842	9321	-148	13	10684	9258	25	6	5811	5862	-89	12	2124	2248	-170					
0	6	21758	19158		7	15478	17911	80	8	7072	5647	-143	14	19574	17393	96	15	3575	3031	105					
2		18592	20073	57	8	7512	7575	127	9	5146	4586	12	15	3575	3031	105	2	8	6607	4803	180				
4		7568	8825	-44	9	8148	9056	147	10	12799	12098	-108	7	8054	11017	55	10	14392	13221	-128					
6		19860	15097	-54	10	8461	8726	-176	8	2845	1911	-91	9	6882	6859	45	11	5591	2118	150					
8		19650	21911	11	11	1799	2089	-85	9	9705	9458	-26	10	6705	5773	140	12	14751	13235	-74					
10		6567	5847	32	12	6814	6602	-166	11	6535	5846	-10	11	15290	17173	96	13	4638	3789	95					
12		5954	6924	-64	13	5701	5989	50	12	6531	5614	100	12	6546	6355	-140	14	12525	11507	39					
14		7884	10252	0	14	1381	1502	-155	13	5634	6191	35	13	6965	6548	135	15	15554	14619	-45					
0	7	5358	5248	-52	15	6848	7697	101	14	2701	5860	-154	14	2886	3359	-17	6	8087	6591	81					
2		7568	12652	-55	1	2	5685	5591	178	15	10901	9495	-68	2	1	6289	6538	0	8	6112	5568	-122			
4		9960	9796	4	3	2545	2563	178	16	15250	11415	9	3	4675	5927	-151	9	8461	7866	-45					
6		8885	7818	-123	4	12416	12105	-174	17	5158	2779	-79	6	2351	1995	151	10	5951	2962	-45					
8		3558	2002	155	5	8204	7412	-57	18	5670	4715	23	7	9694	10172	100	2	8	2868	3267	163				
10		12559	12157	158	10	8657	8400	-10	19	8444	7616	-85	8	5973	6252	-120	11	18820	16753	-61					
12		7788	8712	-140	11	5468	5642	108	20	5752	5247	-98	11	17470	17415	75	12	7435	6142	-70					
14		6248	5110	180	12	5650	5065	-27	21	5228	2296	16	13	3575	2925	75	13	4460	5755	-103					
0	9	14719	14123	150	13	5629	6135	-129	9	6555	5986	-166	14	5282	5116	150	14	10205	8511	-80					
2		8022	8422	-149	14	2601	2571	45	15	2055	1769	88	8	6251	6012	134	15	5895	4483	86					
4		12143	12657	-157	15	6928	8258	-75	16	4049	3720	150	2	10703	10357	-61	10	5023	2208	120					
6		2994	2190	59	16	10564	14427	0	17	4859	5198	-84	1	6407	6750	21	11	2906	2270	-106					
8		6856	6970	-171	17	9622	8359	-168	18	798	1402	-93	10	5866	5445	-23	12	7206	6366	-122					
10		5112	6187	-155	18	8211	7871	-89	19	6577	7847	108	11	8313	7781	-110	13	4537	5815	84</td					

Table 1 (cont.)

$h$	$k$	$l$	$F_o$	$F_c$	$\alpha_{hk\bar{l}}$	$h$	$k$	$l$	$F_o$	$F_c$	$\alpha_{hk\bar{l}}$	$h$	$k$	$l$	$F_o$	$F_c$	$\alpha_{hk\bar{l}}$	$h$	$k$	$l$	$F_o$	$F_c$	$\alpha_{hk\bar{l}}$		
2	12	11	7194	7889	97	5	4	8	7158	6956	-127	5	13	8	8700	8759	-161	4	4	10	6559	6448	15		
12		3349	3415	-17		10	11	11147	69		9	2252	1821	-78	12	4002	4227	22	4	15	0	6982	6597	-147	
2	13	0	5252	5422	0	10	8833	8800	-16	10	2259	3500	-66	13	2429	2418	41	4	15	0	1555	1556	-164		
1		6856	6828	79		11	6892	6594	96	11	2319	2358	-77	14	2305	2598	12	5	20	0	889	889	-19		
2		5975	2199	158		12	9391	9743	26	3	14	9394	8465	-180	4	5	0	11890	10568	180	7	6173	5812	1	
3		11155	10864	72		13	4234	5078	143	1	7085	5893	-76	1	12663	12830	151	7	6173	5812	94				
4		6874	6064	-166		14	2344	2872	180	2	4492	4027	-180	2	4887	4170	117	6	2522	2296	1				
5		8559	7249	125		5	5	0	14089	12556	0	3	10312	9524	-87	3	11973	11749	-132	9	4619	5515	164		
6		2612	1924	-35		1	13358	12412	95	4	4220	3148	171	4	7814	7505	167	4	15	0	9365	8891	0		
7		4492	3288	109		2	13910	13485	44	5	6502	5477	-96	5	13151	12816	-19	1		3567	2429	41			
8		2294	1213	38		3	5726	4943	96	6	4355	3749	177	6	4388	3409	-134	2		6439	5449	8			
9		5598	5132	70		4	6928	6124	-87	7	6577	6065	-78	7	6621	5739	75	3		3315	2857	52			
10		3988	3866	145		5	12493	11670	103	8	3787	3526	159	8	4779	4183	166	4		6677	6030	4			
11		5665	5765	85		6	9387	9349	-40	9	5050	4765	-74	9	6823	6323	-177	5		5361	4957	131			
2	14	0	5518	4569	180	7	10460	10510	82	10	2763	2654	-115	10	2191	977	-148	6		5197	4576	1			
1		5484	5072	129		8	8665	8009	59	3	15	0	2499	1467	0	11	6797	6934	-77	7		2177	1889	37	
2		6844	6246	-83		9	7226	6998	127	1	2476	2910	-177	12	2096	1980	174	4	16	0	6634	6259	0		
3		7566	6100	-99		10	3671	3432	64	2	6844	6494	93	13	2355	3083	14	4	16	0	2794	2574	-39		
4		5983	4775	-12		11	4839	4805	112	3	5572	4898	-83	14	1878	2105	-135	2		4718	3651	17			
5		5766	4472	-47		12	5241	5929	77	4	8894	8580	-171	4	6	0	15809	14403	-180	4		5895	6087	2	
6		6603	5661	17		13	3429	5155	-12	5	3602	3589	-74	1	14376	13586	-43	3		3975	3592	135			
7		4845	3257	70		14	3525	5155	-20	6	6102	6182	-69	2	8770	7293	-164	4		4081	3793	5			
8		3518	3511	-148		15	11049	9461	0	7	1240	2298	66	3	5578	5164	41	5		3108	3312	-108			
9		1510	1896	-172		16	5789	4806	60	8	1251	4520	-48	4	11077	1089	166	7		3108	3312	-108			
10		4505	5353	-54		17	4328	4078	104	9	2351	1816	-136	5	9248	8892	-144	4	17	0	2405	2591	9		
11		3951	5451	-83		18	6565	6616	168	3	16	0	2835	2292	-180	6	9248	8568	-99	4	17	0	1831	1710	180
2	15	0	7943	6654	0	19	4441	3509	168	1	3013	1989	80	7	7419	6936	-167	1		5229	5121	160			
1		11233	9342	-70		20	9797	9704	-41	2	7672	7053	-47	8	8344	8271	-42	2		1695	1619	-112			
3		5619	5059	-97		21	8088	2860	35	3	5741	5513	72	9	6642	6723	-153	3		5875	6109	-112			
4		2751	2273	133		22	5425	5574	143	4	5736	5247	18	10	4613	4738	109	5	1	1	5174	5550	-138		
5		7870	6970	-89		23	3271	2495	126	5	5010	4865	123	11	5721	6431	-178	6		1649	1559	-80			
6		4158	3198	-25		24	2452	2019	-33	6	2916	2328	66	12	4219	4016	-165	7		15403	15557	-38			
7		8580	8071	-94		25	2344	2051	-130	7	1521	980	97	13	3591	4261	-148	4	18	0	4839	5298	-180		
8		2965	2524	-5		26	2556	3001	-39	8	3315	3205	-142	4	7080	9018	-180	1		2393	2047	-69			
9		4421	4426	-5		27	0	23967	23614	-180	3	17	1	4575	4459	75	1		4992	5459	-165				
2	16	4256	3650	0		28	16514	16228	-70	2	5274	4538	-16	2	6140	5196	-133	4	19	0	4151	4136	-178		
1		2946	1914	-71		29	12181	10586	-50	3	5711	5781	83	3	3280	2337	52	5	20	0	3262	3143	180		
3		2527	5466	82		30	5593	4757	-20	4	5027	5109	10	4	9561	8874	-169	5	1	1	5141	5512	-153		
4		4613	3612	-43		31	7149	3445	59	5	7018	7113	111	5	4845	4617	-71	5	1	1	5174	5550	-161		
5		1983	1426	-133		32	5246	3887	-106	6	2499	2529	41	6	3810	4116	-165	6		15403	15557	-38			
6		5691	5480	-147		33	11492	11219	136	7	5244	5174	94	7	5947	5279	-111	3		8126	7179	-94			
7		7493	7735	-82		34	7201	11675	-81	8	5239	5097	154	8	6694	5898	176	4		21765	21999	20			
8		5263	5779	33		35	10068	9592	-152	9	4441	4772	86	9	10208	9941	72	5	2	0	13138	13756	0		
9		4963	5056	-67		36	7341	7266	-72	10	4607	5263	179	10	4844	6082	0	1		2012	2367	-22			
2	17	0	4017	3704	-180	37	2994	2431	-100	11	4607	5263	179	12	4044	4864	-93	6		13750	14429	-46			
1		6475	5780	144		38	3599	4475	72	12	4718	4664	-141	13	4905	6189	19	7		6229	5928	62			
3		2489	4687	-88		39	2699	3414	-76	13	3427	15136	9	14	6016	6067	-5	8		10622	10371	14			
4		4887	4528	-78		40	5478	4602	129	14	5052	4956	-153	15	4905	5180	75	9		6598	6900	52			
5		2464	2528	-99		41	6546	5830	-93	15	5241	4224	-156	16	6616	5952	151	10		1663	4195	-133			
6		2127	1934	0		42	5101	5027	-90	16	5639	5525	52	17	5146	4738	-132	3		3937	3255	79			
7		2804	2796	49		43	6475	6972	-70	17	5772	5200	-126	18	5639	6343	8	4		4797	4110	-75			
8		3208	3793	113		44	3200	3443	-75	19	5772	5594	-126	19	5639	6404	-4	5		2955	2055	-71			
9		5781	7492	74		45	8822	7530	49	20	5701	6094	-126	21	5489	6065	94	6		2742	1953	-72			
10		14269	14789	-58		46	5061	4521	135	22	7721	7170	167	23	4925	5244	8	7		3623	3080	-74			
11		1518	16703	142		47	3542	5961	-152	24	5566	5568	74	24	5606	5952	151	8		2876	2316	18			
12		9592	8876	-100		48	5709	4798	26	25	4657	4120	162	25	5623	5337	0	9		1280	12386	16			
13		10309	9497	-77		49	3151	3047	-121	26	4613	3923	-125	26	5541	4492	62	10		6555	6396	-108			
14		10844	14115	-83		50	3779	3558	103	27	5567	4917	21	27	5347	5274	8	11		13903	14376	-128			
15		18737	20100	-172		51	9004	9189	-35	28	5701	5231	176	28	6847	7415	133	12		2974	1972	-134			
16		14667	15257	-74		52	5146	5629	94	29	5902	7915	61	29	5656	5425	180	13		2778	2445	103			
17		8753	7769	-73		53	4499	5609	46	30	5746	5744	28	30	5607	5321	104	14		3756	3271	104			
18		4694	3846	-126	</td																				

Table 1 (cont.)

$h$	$k$	$l$	$F_o$	$F_c$	$\alpha_{hkl}$	$h$	$k$	$l$	$F_o$	$F_c$	$\alpha_{hkl}$	$h$	$k$	$l$	$F_o$	$F_c$	$\alpha_{hkl}$	$h$	$k$	$l$	$F_o$	$F_c$	$\alpha_{hkl}$									
5	7	8	5219	4781	-107	6	1	1	7910	8978	-160	6	10	9	3725	4602	8	7	6	9	3631	3624	-8	8	4	3	3418	2982	-67			
9	6	6	4464	4010	-21	3	4479	4447	-41	5101	4170	-86	1	5716	4619	163	7	7	10	1482	1691	180	4	4441	4368	-141	4	4	0	3280	2561	-101
10	8	4	8547	8910	-21	4	3559	2931	-126	5038	5548	-56	2	7466	6648	82	1	11812	12271	115	5	3443	4144	-164	5	5	0	3032	3310	-71		
11	5	6	6767	6372	-32	5	5929	5153	-65	6767	6372	-32	3	6831	6736	-109	2	3409	3212	-99	7	7	10	1560	1579	-95	6	6	0	3032	3310	-71
12	0	17005	15883	0	6	5915	5057	-28	6767	6372	-32	4	6060	5490	-168	3	4492	3960	-170	9	9	10	2204	2456	-170	5	5	0	2204	2456	-170	
1	3032	2289	-92	7	3145	3683	-68	1	5565	4828	-43	2	5562	5609	39	4	5562	5300	31	8	5	0	9912	9757	0	8	4	3	3418	2982	-67	
2	7427	6365	-37	8	5557	5089	-33	6767	6372	-32	6	6475	6288	-46	5	5685	5300	31	1	2191	1652	83	2	2	0	4441	4368	-141				
3	6515	5860	-58	9	3607	3129	-178	7	3358	3187	51	6	3297	2484	135	5	3297	2484	135	2	7218	7349	87	3	3	0	3280	2561	-101			
4	9495	8628	-4	10	1847	189	46	8	4335	4426	32	7	11992	12752	85	4	3443	4144	-164	4	4	0	3032	3310	-71							
5	5366	4479	-142	11	6002	6333	-72	9	2876	3297	-169	9	6303	7259	147	5	3032	3310	-71	4	4	0	3032	3310	-71							
6	8340	7544	-13	13	900	1192	-93	10	3460	2974	116	10	2393	2514	-6	10	2393	2514	-6	4	5101	4692	-146	5	5	0	3032	3310	-71			
7	3686	2943	-84	6	2	4934	5414	0	6	2055	517	180	7	8	2994	2697	0	6	4518	4691	-53	6	6	0	3032	3310	-71					
8	10298	15643	10	1	9842	10127	130	1	9119	8382	-64	8	1	7400	6479	68	8	6294	7417	22	8	5	0	9912	9757	0						
9	3288	2366	-65	2	6466	5710	117	2	6707	5787	-30	2	1878	1848	-25	9	1649	1851	81	10	10	0	2294	3580	115							
10	3208	3215	44	3	2825	1377	-101	3	6126	5283	-67	3	9625	9325	84	10	2294	3580	115	1	1	0	3280	2561	-101							
11	3702	5077	-17	4	8707	7916	-152	4	6936	5950	-9	4	2269	4567	59	1	2191	1652	83	2	2	0	3280	2561	-101							
12	8225	6111	0	5	6847	6063	-89	5	5920	4876	-123	5	7755	7626	112	2	7109	7700	-58	3	3	0	3280	2561	-101							
13	2187	2187	0	6	3141	4146	-75	6	6353	3246	75	6	2495	2421	100	2	4486	4560	102	7	7	0	3032	3310	-71							
14	2470	3245	80	7	7200	6772	69	7	6732	6666	-78	8	2178	2100	-48	8	1249	1240	-44	4	10767	11334	19	5	5	0	3032	3310	-71			
15	6021	5076	173	8	2044	7893	-115	9	4109	3612	-113	9	3796	6400	82	8	2026	2467	68	6	5650	6450	68	10	10	0	2294	3580	115			
16	6376	5815	-17	10	4652	3441	39	10	2896	2577	73	10	10	2026	2467	68	10	1649	1851	81	11	11	0	2294	3580	115						
17	2243	1629	-52	11	2742	4220	59	11	5229	5303	115	11	2026	2467	68	11	1649	1851	81	12	12	0	2294	3580	115							
18	7373	7129	28	12	16364	17997	-180	12	3341	2252	-23	12	4524	4531	-49	12	1649	4052	-24	13	13	0	2294	3580	115							
19	2522	2389	-15	13	1	5696	5068	83	13	5355	5214	3	13	1831	1510	120	13	2440	3393	110	14	14	0	2294	3580	115						
20	3847	3577	108	14	16034	17451	70	14	16034	17451	70	14	3796	6400	82	14	10767	11334	19	15	15	0	2294	3580	115							
21	12907	11292	-180	15	4652	3441	39	15	2699	2577	73	15	1649	1851	81	16	1649	1851	81	17	17	0	2294	3580	115							
22	8532	6896	140	16	1592	12926	115	16	5056	5394	0	16	1800	684	158	16	3032	3310	-71	17	17	0	2294	3580	115							
23	1124	9098	-120	17	5707	7869	157	17	5279	5033	115	17	3748	3748	-120	17	3702	3133	81	18	18	0	2294	3580	115							
24	1124	9254	144	18	5707	6282	-157	18	5257	4961	114	18	2580	2076	14	18	5634	5680	-13	19	19	0	2294	3580	115							
25	1432	10152	141	19	10264	16262	-165	19	4767	4633	-121	19	3825	4545	15	19	3460	2920	98	20	20	0	2294	3580	115							
26	7	4322	1327	68	9	7294	7795	69	4	4926	4926	-165	4	2580	2076	14	4	5634	5680	-13	5	5	0	2294	3580	115						
27	8	10323	10708	-154	10	2896	3110	-71	10	6212	2424	125	10	4634	3959	-177	10	2765	2129	24	11	11	0	2294	3580	115						
28	5061	5021	128	11	7520	7882	87	11	3199	3566	-92	11	3748	2346	-127	11	2544	2733	106	12	12	0	2294	3580	115							
29	5710	1404	-108	12	2204	2182	89	12	5678	4318	79	12	1371	1479	-110	12	1523	1523	99	13	13	0	2294	3580	115							
30	1197	2391	-64	13	3485	5606	136	13	4466	4280	-180	13	2678	2197	51	13	7174	1752	0	14	14	0	2294	3580	115							
31	0	4388	3389	-180	14	6275	5910	-180	14	4193	3866	100	14	8733	8917	-51	14	2654	1729	-75	15	15	0	2294	3580	115						
32	1	3297	2112	153	15	7097	7171	103	15	2226	2823	-153	15	2256	2320	125	15	6961	6731	133	16	16	0	2294	3580	115						
33	10205	9428	172	16	2393	1971	116	16	6050	6502	86	16	2815	2566	55	16	2825	3060	-45	17	17	0	2294	3580	115							
34	5446	4450	52	17	4206	4260	57	17	2634	2593	147	17	4751	5612	-98	17	4165	4072	-81	18	18	0	2294	3580	115							
35	7921	6937	-152	18	4857	4080	158	18	3748	3468	105	18	7188	7601	-38	18	7166	8599	158	19	19	0	2294	3580	115							
36	6588	5175	122	19	3583	2695	-156	19	3944	3577	85	19	3802	3024	-137	19	2613	2922	-79	20	20	0	2294	3580	115							
37	5095	4945	139	20	4248	3881	-162	20	3944	3577	85	20	13478	15277	85	20	1613	1613	-48	21	21	0	2294	3580	115							
38	2522	1202	-180	21	3228	3278	-85	21	3061	3024	-137	21	7052	7371	116	21	3655	4211	171	22	22	0	2294	3580	115							
39	2556	1676	162	22	1700	1786	-124	22	3061	3024	-137	22	7052	7371	116	22	3655	4211	171	23	23	0	2294	3580	115							
40	7431	7045	-51	23	4981	4137	180	23	2645	2471	26	23	3559	3554	164	23	1649	1851	-87	24	24	0	2294	3580	115							
41	5880	5107	-57	24	3659	2805	-19	24	3537	3223	94	24	3288	3223	94	24	3050	2505	99	25	25	0	2294	3580	115							
42	3764	3265	-69	25	3223	3278	-68	25	3521	3224	-25	25	3010	31984	-1	25	3773	3249	79	26	26	0	2294	3580	115							
43	3292	3265	0	26	3435	5165	55	26	3087	3537	91	26	4188	4185	0	26	3591	4882	-65	27	27	0	2294	3580	115							
44	3295	3268	-166	27	3199	2620	0	27	3929	3679	24	27	2256	1717	147	27	3509	2855	83	28	28	0	2294	3580	115							
45	4969	4785	14	28	6182	5100	-160	28	1649	719	0	28	1594	1578	0	28	4002	4822	-43	29	29	0	2294	3580	115							
46	3764	4846	-146	29	3659	3241	-75</																									

Table 1 (cont.)

$h$	$k$	$l$	$F_o$	$F_c$	$\alpha_{hkl}$	$h$	$k$	$l$	$F_o$	$F_c$	$\alpha_{hkl}$	$h$	$k$	$l$	$F_o$	$F_c$	$\alpha_{hkl}$	$h$	$k$	$l$	$F_o$	$F_c$	$\alpha_{hkl}$						
9	5	3	3392	3279	75	9	8	3	2306	2400	84	9	12	0	1939	1542	-180	10	2	6	2440	3257	-29						
4			7502	8584	3	4			5599	4452	178	5			4749	4966	-56	7	1443	5221	-111	1	5382	7274	108				
6			3392	3860	23	5	1272	1268	-165	10	0	1	5875	6837	-180	10	3	1	1649	1427	-135	2	2906	3911	-149				
7			1997	2861	-106	6			5061	4199	176	1			5645	7444	115	2	4473	5197	-27	3	2654	2814	-179				
8			2322	4329	39	7			3710	3865	-180	2			6699	8188	-139	3	5776	7809	-111	5	5623	4705	27				
9	6	2	1397	2107	-32	1			4749	4290	-120	3			5765	4820	165	4	6150	8284	-5	10	7	1	2110	2611	96		
3			1997	2107	-94	2	1397	1214	-120	4			3719	4620	127	5	5152	6704	-45	2	3795	4372	-154						
3			2855	2906	50	3			2804	3169	74	5			4342	5983	9	6	2235	3800	64	3	1882	2558	98				
4			2731	2962	43	4			2151	2502	48	6			3190	4782	144	10	4	1	3689	4046	-4	4	2936	3688	158		
5			2955	2636	157	5			2623	2616	-179	7			4151	6661	88	2	2906	3399	44	10	8	1	5409	6428	-68		
7			1851	2106	-82	6			1381	1297	152	10	1	1	2855	5061	7	3	1613	1897	-141	2	5079	6103	83				
9	7	0	2487	2168	-180	9	10	0	4179	4264	0	2			2916	3121	134	4	2083	1963	0	3	2452	2892	46				
1			3756	4063	119	1			4095	4721	144	3			3271	5929	91	5	2151	2462	-60	10	9	0	2667	2626	0		
2			4981	5550	145	2	4857	5947	33	4			5042	3610	-149	6	1295	2111	-4	1	1649	1799	-134						
3			4524	5147	-135	3	2865	4049	-122	5			2151	2332	115	10	5	0	2601	2308	0	10	10	0	1785	1551	0		
4			1769	9213	-166	4	1666	2397	-54	6			1924	2875	-167	1	2381	2135	-17	11	1	0	4499	5219	0				
5			5578	5694	-20	5	4912	4623	-26	7			798	1399	151	2	5032	5727	133	1	3271	3964	-100						
6			2022	3461	-132	9	11	0	4755	6569	0	10	2		6664	7697	C	3	3315	4348	83	2	4165	5394	87				
7			4028	5098	-39	1			176	1490	0	1			5599	4268	-49	4	2994	3520	-172	11	2	0	3733	4416	-180		
8	0		7321	7946	180	2			5282	5423	-19	2			1799	1314	52	5	2506	5129	156	1	5817	5217	-81				
1			3208	3091	-30	3			760	959	-143	5			1924	2024	56	6	1878	3328	-90	2	2876	3152	-45				
2			4179	4841	-173	4	2623	5176	21	5			2306	5065	-164														

Table 2. Atomic and thermal parameters and their standard deviations ( $\times 10^4$ , in parentheses)

	$x/a$	$y/b$	$z/c$	$U_{150}$
Sb(1)	0.3615 (2)	0.1627 (1)	0.5019 (1)	0.0186 (5)
Sb(2)	0.4355 (2)	0.3199 (1)	0.1377 (2)	0.0188 (5)
K(1)	0.3803 (11)	0.0175 (5)	0.2367 (8)	0.0363 (20)
K(2)	0.3024 (10)	0.5652 (5)	0.3331 (8)	0.0329 (18)
O(1)	0.1998 (28)	0.1040 (15)	0.3827 (23)	0.0264 (56)
O(2)	0.0124 (37)	0.1286 (19)	0.2694 (24)	0.0415 (73)
O(3)	0.2601 (28)	0.2514 (13)	0.1906 (19)	0.0205 (48)
O(4)	0.2477 (28)	0.2614 (12)	0.4364 (18)	0.0156 (43)
O(5)	0.5364 (29)	0.2543 (15)	0.5284 (21)	0.0252 (55)
O(6)	0.4961 (27)	0.1452 (14)	0.3662 (20)	0.0201 (48)
O(7)	0.7625 (35)	0.3019 (16)	0.4621 (23)	0.0332 (61)
O(8)	0.6740 (31)	0.0995 (16)	0.1601 (23)	0.0329 (62)
O(9)	0.5167 (27)	0.1946 (14)	0.0994 (20)	0.0219 (50)
O(10)	0.5838 (30)	0.3056 (16)	0.2510 (23)	0.0280 (57)
O(11)	0.3667 (42)	0.4400 (21)	0.5202 (33)	0.0561 (92)
O(12)	0.3303 (27)	0.4064 (15)	0.2478 (21)	0.0219 (51)
O(13)	0.1094 (34)	0.4281 (18)	0.3304 (26)	0.0373 (69)
O(14)	0.6728 (35)	0.4344 (19)	0.5993 (26)	0.0395 (73)
O(15)	0.4679 (52)	-0.0083 (30)	0.4927 (43)	0.1019 (124)
C(1)	0.1083 (40)	0.1543 (21)	0.3307 (25)	0.0123 (57)
C(2)	0.1294 (35)	0.2473 (19)	0.3618 (30)	0.0216 (69)
C(3)	0.1548 (35)	0.2929 (21)	0.2544 (28)	0.0156 (63)
C(4)	0.1946 (40)	0.3797 (19)	0.2833 (27)	0.0139 (59)
C(5)	0.6472 (36)	0.2506 (20)	0.4586 (30)	0.0217 (71)
C(6)	0.6376 (37)	0.1880 (19)	0.3673 (28)	0.0157 (61)
C(7)	0.6644 (43)	0.2270 (20)	0.2570 (29)	0.0186 (67)
C(8)	0.6169 (34)	0.1697 (22)	0.1689 (31)	0.0267 (77)

cess was continued using the full-matrix least-squares method assuming independent isotropic thermal parameters for each atom. The weighting scheme used was

$$\nu\omega = \sqrt{\left( \frac{1}{1 + \left( \frac{|F_o| - b}{a} \right)^2} \right)},$$

where  $a=b=40$  on the same scale as Table 1. The final reliability index for the observed reflexions was  $R=0.11$ . Table 1 lists the observed structure amplitudes and calculated structure factors based on the final atomic coordinates given in Table 2. Atomic scattering factors used were those in *International Tables for X-ray Crystallography* (1962).

### Description and discussion of the structure

It follows from the interatomic distances and angles given in Tables 3 and 4 as well as from Figs. 1 and 2 that the crystal structure of 'tartar emetic' is built up of dimeric tartratoantimonate(III) ions,  $[\text{Sb}_2(\text{C}_4\text{H}_2\text{O}_6)]^{2-}$ . It confirms the previously reported structure of this ion in ammonium antimonyl-( $\pm$ )-tartrate monohydrate (Kiosse, Golovastikov & Belov, 1964), tris-*o*-phenanthroline-iron(II) antimony(+)-tartrate (Templeton, Zalkin & Ueki, 1966) and ammonium antimonyl-(+)-tartrate hemihydrate (Kiosse, Golovastikov, Ablov & Belov, 1967). Both antimony atoms in the  $[\text{Sb}_2(\text{C}_4\text{H}_2\text{O}_6)]^{2-}$  ion are bound to four oxygen atoms in a slightly deformed square pyramid whose apex is

occupied by the unshared electron pair. Two oxygen atoms are from two carboxyl groups and the other two oxygen atoms from the  $\alpha$ -hydroxyl groups at mean distances of 2.20 and 2.01 Å respectively. These values of the Sb-O bond lengths are close to those found in  $(\text{NH}_4)_2[\text{Sb}_2(\text{D-tart})_2] \cdot 3\text{H}_2\text{O}$  (2.18 and 2.02 Å respectively) as well as in  $(\text{NH}_4)_2[\text{Sb}_2(\text{D,L-tart})_2] \cdot 4\text{H}_2\text{O}$

(2.15 and 2.04 Å respectively). The antimony atom thus belongs to two nearly planar five-membered rings with an average O-Sb-O angle of 80°.

The anion  $[\text{Sb}_2(\text{C}_4\text{H}_2\text{O}_6)]^{2-}$  is asymmetric in the crystal structure, since the space group  $Pca2_1$  does not require any point symmetry, but, if idealized, the anion would have point symmetry  $D_2$ . Racemic 'tartar emetic'

Table 3. Some interatomic distances with standard deviations (in parentheses)

The positions are denoted as follows:			
No superscript	x	y	z
(i)	$\frac{1}{2} + x$ ,	$-y$	$z$
(ii)	$-x$ ,	$-y$ ,	$\frac{1}{2} + z$
<i>(a)</i> Within the tartratoantimonate(III) ion			
Sb(1)-O(1)	2.25 (0.03) Å		
Sb(1)-O(4)	2.06 (0.03)		
Sb(1)-O(5)	2.17 (0.03)		
Sb(1)-O(6)	2.05 (0.02)		
Sb(2)-O(3)	2.01 (0.03)		
Sb(2)-O(9)	2.22 (0.03)		
Sb(2)-O(10)	1.91 (0.03)		
Sb(2)-O(12)	2.16 (0.03)		
O(1)-C(1)	1.31 (0.04)		
O(2)-C(1)	1.20 (0.05)		
O(4)-C(2)	1.40 (0.04)		
O(5)-C(5)	1.29 (0.04)		
O(6)-C(6)	1.43 (0.04)		
O(7)-C(5)	1.31 (0.05)		
O(3)-C(3)	1.39 (0.04)		
O(9)-C(8)	1.29 (0.04)		
O(8)-C(8)	1.26 (0.04)		
O(10)-C(7)	1.47 (0.05)		
O(12)-C(4)	1.34 (0.04)		
O(13)-C(4)	1.23 (0.04)		
C(1)-C(2)	1.57 (0.05)		
C(2)-C(3)	1.52 (0.05)		
C(3)-C(4)	1.50 (0.05)		
C(5)-C(6)	1.51 (0.05)		
C(6)-C(7)	1.51 (0.05)		
C(7)-C(8)	1.48 (0.05)		
<i>(b)</i> Coordination about K(1)			
K(1)···O(1)	2.77 (0.04) Å		
K(1)···O(2 <sup>ii</sup> )	2.68 (0.04)		
K(1)···O(6)	2.81 (0.04)		
K(1)···O(8 <sup>i</sup> )	2.79 (0.04)		
K(1)···O(8)	3.06 (0.04)		
K(1)···O(15 <sup>ii</sup> )	3.26 (0.06)		
<i>(c)</i> Coordination about K(2)			
K(2)···O(7 <sup>i</sup> )	2.70 (0.04) Å		
K(2)···O(10 <sup>i</sup> )	3.02 (0.04)		
K(2)···O(11)	3.11 (0.05)		
K(2)···O(12)	2.80 (0.04)		
K(2)···O(13)	2.81 (0.04)		
K(2)···O(13 <sup>i</sup> )	2.70 (0.04)		
K(2)···O(14 <sup>ii</sup> )	2.86 (0.04)		
<i>(d)</i> Hydrogen-bonded contact distances			
O(11)···O(14)	2.86 Å		
O(11)···O(4)	3.26		
O(14)···O(11 <sup>i</sup> )	2.84		
O(14)···O(7)	2.85		
O(15)···O(6)	2.95		
O(15)···O(8 <sup>ii</sup> )	2.82		

Table 4. Some interatomic angles with standard deviations (in parentheses)

<i>(a)</i> At the antimony atoms			
O(1)-Sb(1)-O(4)	77.0 (1.0)°	O(3)-Sb(2)-O(9)	78.5 (1.0)
O(1)-Sb(1)-O(6)	77.5 (1.0)	O(3)-Sb(2)-O(12)	80.6 (1.0)
O(4)-Sb(1)-O(5)	82.0 (1.0)	O(9)-Sb(2)-O(10)	79.7 (1.0)
O(5)-Sb(1)-O(6)	78.9 (1.0)	O(10)-Sb(2)-O(12)	85.6 (1.0)
<i>(b)</i> Within the tartrate groups			
Sb(1)-O(1)-C(1)	115 (2.3)°	Sb(1)-O(5)-C(5)	113 (2.2)°
Sb(1)-O(4)-C(2)	119 (2.1)	Sb(1)-O(6)-C(6)	115 (2.0)
O(1)-C(1)-C(2)	114 (3.0)	O(5)-C(5)-C(6)	118 (3.1)
C(1)-C(2)-O(4)	113 (2.9)	C(5)-C(6)-O(6)	112 (2.8)
O(1)-C(1)-O(2)	120 (3.4)	O(5)-C(5)-O(7)	121 (3.3)
O(2)-C(1)-C(2)	124 (3.3)	O(7)-C(5)-C(6)	119 (3.2)
C(1)-C(2)-C(3)	106 (2.8)	C(5)-C(6)-C(7)	111 (2.9)
O(4)-C(2)-C(3)	111 (2.9)	O(6)-C(6)-C(7)	109 (2.8)
C(2)-C(3)-O(3)	110 (2.8)	C(6)-C(7)-O(10)	109 (2.9)
C(2)-C(3)-C(4)	107 (2.8)	C(6)-C(7)-C(8)	109 (3.1)
O(3)-C(3)-C(4)	116 (2.9)	O(10)-C(7)-C(8)	112 (3.0)
C(3)-C(4)-O(12)	115 (3.0)	C(7)-C(8)-O(9)	118 (3.3)
C(3)-C(4)-O(13)	125 (3.3)	C(7)-C(8)-O(8)	121 (3.4)
O(12)-C(4)-O(13)	118 (3.3)	O(9)-C(8)-O(8)	120 (3.4)
Sb(2)-O(3)-C(3)	114 (2.0)	Sb(2)-O(10)-C(7)	118 (1.5)
Sb(2)-O(12)-C(4)	111 (2.1)	Sb(2)-O(9)-C(8)	111 (2.2)

is therefore a racemic mixture of dimeric (+)-tartrato-antimonate(III) and (-)-tartratoantimonate(III), and not a ( $\pm$ )-tartratoantimonate whose dimeric anion would be formed of both (+) and (-)-tartaric acid. The (+), (-) combination has been observed so far only in the case of the divanadyl(IV) tartrate ion (Tapsott, Belford & Paul, 1968) but the divanadyl(IV) tartrate ion occurs in the optically active (+) form (Forrest & Prout, 1967). A *meso*-tartrate of this kind is sterically impossible (Reihlen & Hezel, 1931). There is no antimony group in the structure of the so called

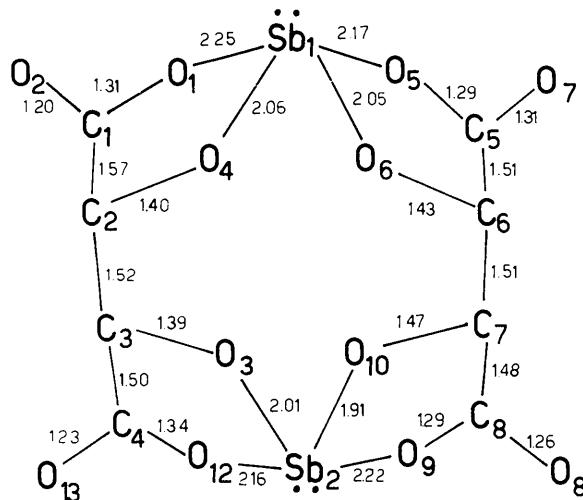


Fig. 1. The  $[Sb_2(C_4H_2O_6)_2]^{2-}$  ion with atom designations and bond lengths.

antimonyl-tartrates; the oxygen atoms required by the chemical analysis belong to the water of crystallization. The proper formula of 'tartar emetic' is thus  $K_2[Sb_2(C_4H_2O_6)_2] \cdot 3H_2O$  and not  $KSbOC_4H_4O_6 \cdot \frac{1}{2}H_2O$  or  $K[Sb(C_4H_2O_6)H_2O] \cdot \frac{1}{2}H_2O$  (*Gmelin's Handbuch*, 1953).

The three crystallographically independent water molecules are differently bound in the crystal structure. While  $H_2O(11)$  and  $H_2O(14)$  build up the mutually hydrogen-bonded zigzag chains between  $[Sb_2(C_4H_2O_6)_2]^{2-}$  ions along the  $a$  axis, the third water molecule [ $H_2O(15)$ ] has no contact with other water molecules. At the same time  $H_2O(11)$  and  $H_2O(14)$  are more closely bound to the potassium ions than  $H_2O(15)$ . The potassium-oxygen(11) and potassium-oxygen(14) distances are 3.11 and 2.86 Å respectively, while the potassium-oxygen(15) distance is 3.26 Å. These results explain the observations that potassium antimonyl tartrate on heating loses one third of its water at 100 °C, while the remaining two thirds is driven off only at 200 to 220 °C (*Gmelin's Handbuch*, 1953). This is probably also the reason for the relatively high temperature factor found for the oxygen atom which belongs to the water molecule  $H_2O(15)$ . The hydrogen bond lengths are given in Table 3.

The coordination of each potassium ion is different both as to the number of and the distances from the surrounding oxygen atoms. There are six oxygen atoms about K(1) at distances of 2.68 to 3.26 Å and seven-fold coordination is completed by the unshared electron pair from the antimony atom Sb(1). There are seven oxygen atoms about K(2) at distances of 2.70 to 3.11 Å and eightfold coordination is similarly completed

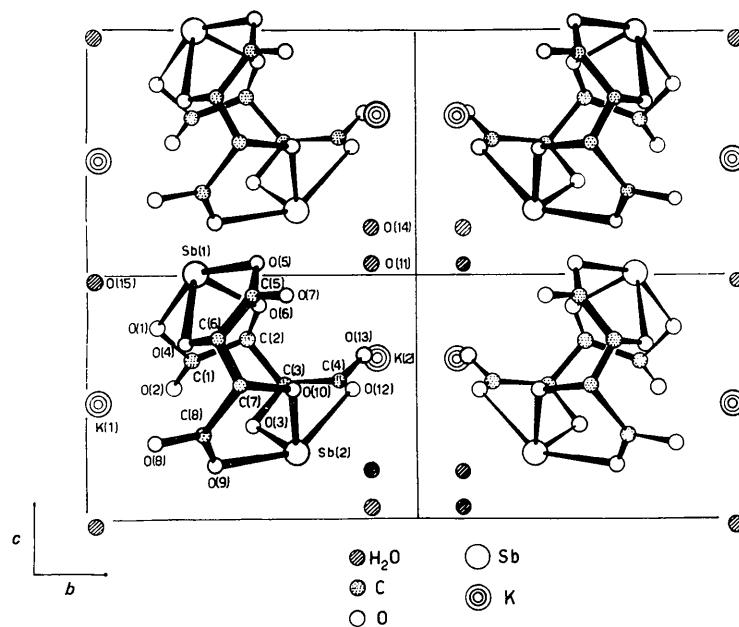


Fig. 2. The crystal structure of racemic potassium di- $\mu$ -tartratodianimonate(III) trihydrate projected along the  $a$  axis.

by the unshared electron pair from Sb(2). The next nearest oxygen atom to K(1) is O(15) at 3.46 Å, and symmetrically related O(14) to K(2) at 3.44 Å. The coordination polyhedra about the potassium ions are similar to those found in other potassium salts where the potassium-oxygen distances are also in the range 2.62 to 3.20 Å (Palenik, 1967).

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## X-ray Structure Determination of (+)-Dibromodehydrotetrahydrorugulosin, a Heavy Atom Derivative of (+)-Rugulosin

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The crystal structure of (+)-dibromodehydrotetrahydrorugulosin water and methanol solvate,  $C_{30}H_{22}O_{10}Br_2 \cdot H_2O \cdot 2CH_3OH$ , has been determined in order to elucidate the molecular structure and absolute configuration of rugulosin,  $C_{30}H_{22}O_{10}$ , a fungal pigment isolated from *Penicillium rugulosum* Thom. The crystals are monoclinic with space group  $P2_1$  and the unit-cell dimensions are  $a=9.78$ ,  $b=17.04$ ,  $c=9.45$  Å and  $\beta=98.0^\circ$ . Two formula units are contained in the cell. The crystal structure was solved by the heavy-atom method and refined by the block-matrix least-squares method including anisotropic thermal parameters. The final  $R$  value for 1482 non-zero observed structure factors was 0.109. The absolute configuration was determined by the use of the anomalous dispersion of bromine atoms for Cu  $K\alpha$  radiation.

The molecule consists of two tricyclic rings of partially hydrogenated anthraquinone cross-linked at the  $A$  and  $A'$  rings by means of four C-C covalent bonds. On the basis of the results obtained by the present structure determination, the structures and stereochemistry of (+)-rugulosin, (-)-luteoskyrin and (-)-rubroskyrin have been established, including their absolute configurations.

#### Introduction

(+)-Rugulosin,  $C_{30}H_{22}O_{10}$ , is a fungal pigment isolated from *Penicillium rugulosum* Thom and some other fungi (Shibata, Tanaka, Chihara & Mitsuhashi, 1952; Breen, Dacre, Raistrick & Smith, 1955; Shibata & Udagawa, 1963). The chemical structure of rugulosin has since been extensively investigated and it has been shown that it is a representative of a new group of colouring matters having a dimeric structure of partially hydrogenated anthraquinones (Shibata, Murakami, Kitagawa & Kishi, 1956; Shibata, Murakami & Takido, 1956; Briggs & LeQuesne, 1965; Shibata, Ogihara, Kobayashi, Seo & Kitagawa, 1968; Sankawa, Seo, Kobayashi, Ogihara & Shibata, 1968).

Among the related compounds obtained from *Penicillium islandicum* Sopp, (-)-luteoskyrin,  $C_{30}H_{22}O_{12}$

and rubroskyrin,  $C_{30}H_{22}O_{12}$ , were shown to have very close structures to rugulosin. Luteoskyrin was particularly noted as a toxic principle causing liver damage and sometimes liver cancer in experimental animals. To date, several structures have been put forward for these substances mainly on the basis of chemical and spectroscopic data, but they are not conclusive. An X-ray structure analysis of a heavy-atom derivative of rugulosin was therefore undertaken. A preliminary report of the present study has already been published (Kobayashi, Iitaka, Sankawa, Ogihara & Shibata, 1968).

#### Experimental

Various attempts at bromination of the skeleton of rugulosin failed to produce a suitable derivative for X-ray analysis. It was finally found that only the hydro-